

## 1 Boltzmann probabilities

Consider a three-state system with energies  $(-\epsilon, 0, \epsilon)$ .

- (a) At infinite temperature, what are the probabilities of the three states being occupied? What is the internal energy  $U$ ? What is the entropy  $S$ ?
- (b) At very low temperature, what are the three probabilities?
- (c) What are the three probabilities at zero temperature? What is the internal energy  $U$ ? What is the entropy  $S$ ?
- (d) What happens to the probabilities if you allow the temperature to be negative?